

Date: 13/02/2023

Course:

Second Year B. Pharmacy

Subject Name:

Physical Pharmaceutics I

Max Marks:

75

Sem:

III

Subject Code:

BP302T

Duration:

3 Hr.

Instructions:

1. All questions are compulsory
2. Draw diagrams / figures wherever necessary
3. Figures to right indicate full marks

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Q. 1. Objective Type Questions (Answer all the questions) (10 x 2) = 20

- i) Give any four ways to express solubility of a drug.
- ii) Define critical solution temperature. Give its applications.
- iii) Define – desublimation, polymorphism, vapour pressure, latent heat
- iv) Differentiate crystalline and amorphous solid.
- v) Why interfacial tension is less than surface tension?
- vi) Explain HLB scale.
- vii) Give pharmaceutical applications of complexation.
- viii) Give importance of protein binding.
- ix) Write buffer equation and buffer capacity.
- x) Define isotonic solution and paratonic solution.

Q. 2. Long Answers (Answer 2 out of 3) (2 x 10) = 20

- i) Write in detail about Raoult's law with help of following point,
a) statement of law, b) ideal solution and real solution,
c) positive deviation d) negative deviation
- ii) Describe refractive index property of drug molecule. Explain various refractometers used to determine refractive index in detail.
- iii) Describe in detail methods of analysis of complex.

Q. 3. Short Answers (Answer 7 out of 9) (7 x 5) = 35

- i) Explain in detail various factors affecting solubility of gases in liquid.
- ii) What is liquid crystal? Write its classification with properties of it. Give its applications.
- iii) Define aerosol dosage form. Give its merit and demerits. Explain propellant used in aerosol.
- iv) Explain spreading coefficient along with applications.
- v) Describe in detail about capillary rise method for determination of surface