

RRB PHARMACIST
2024
MODEL PAPER -89
40 QUESTIONS
WITH DETAILED EXPLANATION
SUBJECT -
PHARMACEUTICAL CHEMISTRY

PHARMACY INDIA

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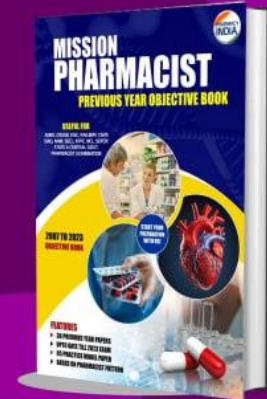
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1. The process in which a solid gets directly converted into vapour is called

- (a) Desiccation
- (b) Sublimation
- (c) Evaporation
- (d) Condensation





1. The process in which a solid gets directly converted into vapour is called

- (a) Desiccation
- (b) Sublimation**
- (c) Evaporation
- (d) Condensation





2 When $\text{pH}=\text{Pka}$, the ratio of conjugate base to acid is

- (a) 1:100
- (b) 1:1000
- (c) 1:10
- (d) 1:1





2 When $\text{pH}=\text{Pka}$, the ratio of conjugate base to acid is

- (a) 1:100
- (b) 1:1000
- (c) 1:10
- (d) 1:1





3. The distribution of probability density calculated from Maxwell Boltzmann statistics in an arbitrarily chosen direction spread with

- (a) Increasing temperature
- (b) Decreasing temperature
- (c) Temperature is independent
- (d) None of these





3. The distribution of probability density calculated from Maxwell Boltzmann statistics in an arbitrarily chosen direction spread with

- (a) Increasing temperature
- (b) Decreasing temperature
- (c) Temperature is independent
- (d) None of these





4. Mass number

- (a) Number of electrons + number of neutrons
- (b) Number of electrons
- (c) Number of protons + number of neutrons
- (d) Number of protons





4. Mass number

- (a) Number of electrons + number of neutrons
- (b) Number of electrons
- (c) Number of protons + number of neutrons
- (d) Number of protons



5. An electron volt is equal to

- (a) 4.4186×10^{14} Hz
- (b) 1.5×10^{-6} m
- (c) 1.602×10^{-12} erg
- (d) 1.5×10 erg





5. An electron volt is equal to

(a) 4.4186×10^{14} Hz

(b) 1.5×10^{-6} m

(c) 1.602×10^{-12} erg

(d) 1.5×10 erg





6. Which among the following exist as a liquid

- (a) Chloral hydrate
- (b) Methyl prylon
- (c) Paraldehyde
- (d) Glutethimide





6. Which among the following exist as a liquid

- (a) Chloral hydrate
- (b) Methyl prylon
- (c) Paraldehyde
- (d) Glutethimide





7. Quantities of electrolytes administered to patients are expressed by

- (a) Milliequivalent
- (b) Equivalent
- (c) Milligram
- (d) Millilitre





7. Quantities of electrolytes administered to patients are expressed by

- (a) Milliequivalent
- (b) Equivalent
- (c) Milligram
- (d) Millilitre





8. Which one of the following is NOT a unit of concentration

- (a) Molarity
- (b) Enthalpy
- (c) Mass per volume
- (d) Milliequivalent





8. Which one of the following is NOT a unit of concentration

- (a) Molarity
- (b) Enthalpy
- (c) Mass per volume
- (d) Milliequivalent



9. The equation involved in the quantum mechanics approach to molecular modelling is

- (a) Boltzmann equation
- (b) Schrodinger wave equation
- (c) Ernst equation
- (d) Arrhenius equation



9. The equation involved in the quantum mechanics approach to molecular modelling is

- (a) Boltzmann equation
- (b) Schrodinger wave equation
- (c) Ernst equation
- (d) Arrhenius equation





10. Sound cannot travel in

- (a) Air
- (b) Steel
- (c) Water
- (d) Vacuum





10. Sound cannot travel in

- (a) Air
- (b) Steel
- (c) Water
- (d) Vacuum





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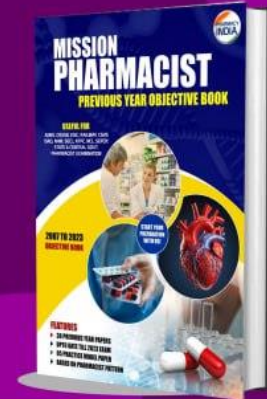
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11. The limit test of Iron depends upon the interaction of iron with

- (a) Barium Sulphate reagent
- (b) Diphenyl thiocarbazon
- (c) Thioglycolic acid
- (d) Mercuric chloride





11. The limit test of Iron depends upon the interaction of iron with

- (a) Barium Sulphate reagent
- (b) Diphenyl thiocarbazon
- (c) Thioglycolic acid
- (d) Mercuric chloride





12. The limit test of lead is done using

- (a) Acetic acid
- (b) Dithizone
- (c) Phenol red
- (d) Thioglycolic acid





12. The limit test of lead is done using

- (a) Acetic acid
- (b) Dithizone
- (c) Phenol red
- (d) Thioglycolic acid





13. The buffer used in the limit test for iron is

- (a) Sodium acetate
- (b) Sodium carbonate
- (c) Ammonia and Ammonium chloride
- (d) Ammonium citrate





13. The buffer used in the limit test for iron is

- (a) Sodium acetate
- (b) Sodium carbonate
- (c) Ammonia and Ammonium chloride
- (d) Ammonium citrate





14. Which one is not the type of heavy metal

- (a) As (Arsenic)
- (b) Hg (Mercury)
- (c) Fe (Iron)
- (d) Pb (Lead)





14. Which one is not the type of heavy metal

- (a) As (Arsenic)
- (b) Hg (Mercury)
- (c) Fe (Iron)
- (d) Pb (Lead)





15. In limit test of chloride, which is compared

- (a) Turbidity
- (b) Opalescence
- (c) Fluorescence
- (d) Colour





15. In limit test of chloride, which is compared

- (a) Turbidity
- (b) Opalescence
- (c) Fluorescence
- (d) Colour





16. The lead acetate cotton wool is used in the limit test for arsenic to

- (a) Make the arsine gas
- (b) Develop the yellow colour
- (c) Trap the HS gas
- (d) None of these





16. The lead acetate cotton wool is used in the limit test for arsenic to

- (a) Make the arsine gas
- (b) Develop the yellow colour
- (c) Trap the HS gas
- (d) None of these





17. Which of the following is not a test for purity

- (a) Specific gravity
- (b) Melting point
- (c) Boiling point
- (d) Quantity of sample





17. Which of the following is not a test for purity

- (a) Specific gravity
- (b) Melting point
- (c) Boiling point
- (d) Quantity of sample





18. Diphenyl thiocarbazono test is a specific test for the determination of

- (a) Arsenic
- (b) Lead
- (c) Sulphate
- (d) Chloride





18. Diphenyl thiocarbazono test is a specific test for the determination of

- (a) Arsenic
- (b) Lead
- (c) Sulphate
- (d) Chloride





19. In chloride limit test, the reagent added for precipitation is

- (a) Sodium chloride
- (b) Silver nitrate
- (c) Sodium nitrate
- (d) Silver chloride





19. In chloride limit test, the reagent added for precipitation is

- (a) Sodium chloride
- (b) Silver nitrate
- (c) Sodium nitrate
- (d) Silver chloride





20. In heavy metals limit test, the standard prepared by using solution of

- (a) Silver nitrate
- (b) Lead nitrate
- (c) Silver chloride
- (d) Hydrogen chloride





20. In heavy metals limit test, the standard prepared by using solution of

- (a) Silver nitrate
- (b) Lead nitrate**
- (c) Silver chloride
- (d) Hydrogen chloride



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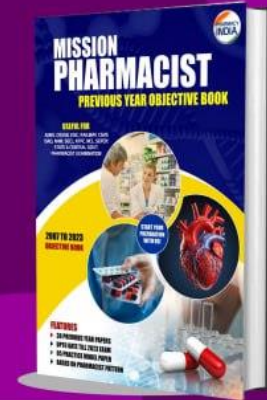
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21. Magnesium sulphate

- (a) Rehydrating agent
- (b) Cathartic agent
- (c) Seizure preventing agent
- (d) Hyperemic agent





21. Magnesium sulphate

- (a) Rehydrating agent
- (b) Cathartic agent
- (c) Seizure preventing agent
- (d) Hyperemic agent





22. Milk of magnesia is an example of which group of laxatives

- (a) Bulk-forming laxative
- (b) Osmotic laxative
- (c) Stimulant laxative
- (d) Faecal softener





22. Milk of magnesia is an example of which group of laxatives

- (a) Bulk-forming laxative
- (b) Osmotic laxative
- (c) Stimulant laxative
- (d) Faecal softener





23. Milk of magnesia is a preparation containing between 7 and 8.5% of

- (a) Magnesium hydroxide
- (b) Magnesium gluconate
- (c) Magnesium perchlorate
- (d) Magnesium oxide





23. Milk of magnesia is a preparation containing between 7 and 8.5% of

- (a) Magnesium hydroxide
- (b) Magnesium gluconate
- (c) Magnesium perchlorate
- (d) Magnesium oxide**





24. Antacids are

- (a) Strong base
- (b) Strong acid
- (c) Weak acids
- (d) Weak base





24. Antacids are

- (a) Strong base
- (b) Strong acid
- (c) Weak acids
- (d) Weak base





25. Bleaching may be caused by antacid preparations containing

- (a) Magnesium carbonate
- (b) Calcium carbonate
- (c) Sodium bicarbonate
- (d) Aluminum hydroxide





25. Bleaching may be caused by antacid preparations containing

- (a) Magnesium carbonate
- (b) Calcium carbonate
- (c) Sodium bicarbonate
- (d) Aluminum hydroxide





26. Chemical formula for talc is

- (a) $4\text{MgO} \cdot 4\text{SiO}_2 \cdot \text{OH}$
- (b) $3\text{MgO} \cdot 3\text{SiO}_2 \cdot \text{OH}$
- (c) $4\text{MgO} \cdot 3\text{SiO}_2 \cdot \text{OH}$
- (d) $\text{Mg}_3\text{Si}_4\text{O}_{10}(\text{OH})_2$





26. Chemical formula for talc is

- (a) $4\text{MgO} \cdot 4\text{SiO}_2 \cdot \text{OH}$
- (b) $3\text{MgO} \cdot 3\text{SiO}_2 \cdot \text{OH}$
- (c) $4\text{MgO} \cdot 3\text{SiO}_2 \cdot \text{OH}$
- (d) $\text{Mg}_3\text{Si}_4\text{O}_{10}(\text{OH})_2$





27. Povidone iodine contains ____ of iodine

- (a) 2.5%
- (b) 2 to 5%
- (c) 5 to 9%
- (d) 9 to 12%





27. Povidone iodine contains ____ of iodine

- (a) 2.5%
- (b) 2 to 5%
- (c) 5 to 9%
- (d) 9 to 12%





28. Sublimed Sulphur is use

- (a) Anti-sebaceous
- (b) Anti-dandruff
- (c) Anti-scabies
- (d) Anti-alopecia





28. Sublimed Sulphur is use

- (a) Anti-sebaceous
- (b) Anti-dandruff
- (c) Anti-scabies
- (d) Anti-alopecia



29. Reddish brown colour of calamine is due to

- (a) Zinc chloride
- (b) Ferric chloride
- (c) Ferric oxide
- (d) Potassium permanganate





29. Reddish brown colour of calamine is due to

- (a) Zinc chloride
- (b) Ferric chloride
- (c) Ferric oxide
- (d) Potassium permanganate





30. Medicated Powders commonly contain

- (a) Iodine
- (b) Talc
- (c) Bentonite
- (d) Cellulose





30. Medicated Powders commonly contain

- (a) Iodine
- (b) Talc
- (c) Bentonite
- (d) Cellulose



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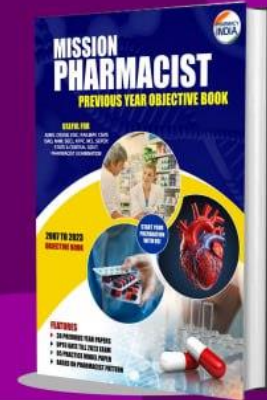
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31. Locally applied protein precipitants with low cell permeability are known as

- (a) Epilatories
- (b) Astringents
- (c) Humectants
- (d) Keratolytics





31. Locally applied protein precipitants with low cell permeability are known as

- (a) Epilatories
- (b) Astringents
- (c) Humectants
- (d) Keratolytics





32. Aluminium ion in solution is an example for

- (a) Germicide
- (b) Astringent
- (c) Protective
- (d) Pigment





32. Aluminium ion in solution is an example for

- (a) Germicide
- (b) Astringent
- (c) Protective
- (d) Pigment



33. External application of dried aluminium hydroxide gel is

- (a) Flavouring agent
- (b) Mild astringent
- (c) Absorbent
- (d) Coloring agent





33. External application of dried aluminium hydroxide gel is

- (a) Flavouring agent
- (b) Mild astringent
- (c) Absorbent
- (d) Coloring agent





34. Bismuth sub carbonate is a/an

- (a) Antacid and laxative
- (b) Adsorbent and astringent
- (c) Saline laxative
- (d) Diuretic and electrolyte replenished





34. Bismuth sub carbonate is a/an

- (a) Antacid and laxative
- (b) Adsorbent and astringent
- (c) Saline laxative
- (d) Diuretic and electrolyte replenished





35. Which of the following is an example of astringent

- (a) Milk
- (b) Tannins
- (c) Fatty acids
- (d) Calcium carbonate





35. Which of the following is an example of astringent

- (a) Milk
- (b) Tannins
- (c) Fatty acids
- (d) Calcium carbonate





36. The mechanism by which fluorides inhibit dental caries to

- (a) By increasing susceptibility to acid
- (b) increasing the sensitivity of tooth
- (c) Decreased acid solubility of enamel
- (d) Increased acid solubility of enamel



36. The mechanism by which fluorides inhibit dental caries to

- (a) By increasing susceptibility to acid
- (b) increasing the sensitivity of tooth
- (c) Decreased acid solubility of enamel
- (d) Increased acid solubility of enamel





37. Stannous fluoride is used as

- (a) Astringent
- (b) Dental caries
- (c) Dentifrices
- (d) Anti-dandruff





37. Stannous fluoride is used as

- (a) Astringent
- (b) Dental caries
- (c) Dentifrices
- (d) Anti-dandruff





38. Abrasive agent used in dentifrices

- (a) Sodium fluoride
- (b) Selenium sulphide
- (c) Zinc sulphate
- (d) Calcium carbonate





38. Abrasive agent used in dentifrices

- (a) Sodium fluoride
- (b) Selenium sulphide
- (c) Zinc sulphate
- (d) Calcium carbonate





39. Sodium fluoride is used to cure

- (a) Inflammation of mouth
- (b) Bacterial infection
- (c) Acidity
- (d) Dental caries





39. Sodium fluoride is used to cure

- (a) Inflammation of mouth
- (b) Bacterial infection
- (c) Acidity
- (d) Dental caries





40. Sodium metaphosphate is used in

- (a) Anticaries
- (b) Antacid
- (c) Desensitizer
- (d) Dentifrice





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