

GATE-1995

PART - A Section-I (Choose The Correct Answer)

1. Multiple choice question

i. One of the drugs is excreted primarily by renal tubular sections. Identity
(a) Gentamycin (b) Kanamycin
(c) Tetracycline (d) Carbenicillin

ii. Identify one of the cancer chemotherapeutic agents which is an antimetabolite
(a) Fluorouracil (b) Nitrogen mustard
(c) Cyclophosphamide (d) Chlorambucil

iii. Agents useful in the treatment of bronchial asthma usually
(a) Block both alpha and beta adrenergic receptors
(b) Stimulate alpha receptors but block beta receptors
(c) Stimulate beta receptors but block alpha receptors
(d) Stimulate alpha and/or beta receptors

iv. The antiarrhythmic drug Quinidine is
(a) (+) Stereoisomer of Quinine
(b) (-) Stereoisomer of Quinine
(c) (+) racemic mixture of Quinine
(d) None of these

v. Beta-Carboline ring system is present in
(a) Emetine alkaloid (b) Cortisone acetate
(c) Deserpidine molecule (d) Atropine alkaloid

vi. Licence for wholesale of drugs specified in schedule C and C, are issued in form
(a) 20 A (b) 20 B (c) 21 B (d) 22

vii. The principal constituents Anethole (50-60%) and Fenchone (18-20%) are present in the volatile oil obtained from
(a) Fruits of Ammi visnaga Linn
(b) Fruits of Foeniculum capillaceum
(c) Fruits of Carum carvi Linn G (F. vulgare)
(d) Fruits of Anethum graveolens Linn

viii. Caffeine on oxidation with $KClO_4/HCl$ gives
(a) Trimethyl alloxan and urea
(b) Methyl alloxan and dimethyl urea
(c) 1, 3 Dimethyl alloxan and mono methyl urea
(d) None of these

ix. One of the following types of adverse drug reactions is not believed to be dose related
ix. Phenomena
(a) Side effects and toxic reactions
(b) Toxic reactions and hypersensitivity
(c) Side effects and hypersensitivity
(d) Hypersensitivity and idiosyncracy

x. Green bones are used for the preparation of a gelatin of the type
(a) A (b) B (c) C (d) A and B

xi. In capsules, ROTOFIL is used for filling M
(a) Powders (b) Pellets
(c) Liquids (d) Corrosive liquids

xii. Gelatin and glycerin are used for the preparation of lamellae in a specified ratio, Identify the correct ratio
(a) 1:1 (b) 5:1 (c) 9:1 (d) 10:1

xiii. The Shick test is used to determine susceptibility to
(a) Measles (b) Diphtheria
(c) Polio (d) Typhoid

xiv. One of the following general characteristics is not true for alkaloids
(a) Nitrogen in the heterocyclic nucleus
(b) Good solubility in organic solvents
(c) pK_S less than 7
(d) Exhibit optical activity

xv. Silver/Silver chloride electrode consists of
(a) Metallic silver coated with a layer of silver chloride
(b) Polished platinum wire coated with silver chloride
(c) Silver wire dipped in a saturated solution of silver chloride
(d) Two electrodes one of silver and other of silver chloride

xvi. In gel permeation chromatography molecules are separated on the basis of their
(a) Chemical nature (b) Size and shape
(c) Adsorptive properties (d) Partition coefficient

xvii. According to Pauli Exclusion Principle, spins of two electrons in the same orbital are
(a) Parallel to one another
(b) Perpendicular to one another
(c) Opposite to one another
(d) Supporting to one another

xviii. Vaccines and toxoids are precipitated and absorbed on to aluminium hydroxide or other suitable media. This process results in a dosage form that in comparison to fluid forms
(a) More effective orally
(b) Absorbed slowly
(c) Stability is increased
(d) Irritation is lost

xix. In order to produce characteristic pharmacological action, a drug must always
xix.
(a) Reach high blood vessels
(b) Absorbed from GIT readily
(c) Achieve adequate concentration at the site of action
(d) Excrete unchanged in urine

xx. The Wurster process can be used to
 (a) Coat tablets
 (b) Determine the disintegration time
 (c) Gas sterilize parenteral solutions
 (d) Automatic filling of capsules

SECTION-II (Fill In The Blank)

2. Fill in the blanks with proper answers

i. In multistation process, the portions of the head that hold the upper and lower punches are called the upper and lower _____ (1) _____ respectively.

ii. In tablet disintegration test, the wire mesh of the tube at its lowest point is at least mm _____ (2) _____ mm above the bottom of the beaker.

iii. Roller compactor is used in large scale _____ (3) _____ granulation.

iv. Chemical name of Propellant 114 is _____ (4) _____

v. In non-aqueous injection, Sesame oil and _____ (5) _____ oil are used as solvents.

vi. 1-(p-chlorobenzoyl) 5-methoxy-2-methyl indole-3-acetic acid is commonly known as _____ (6) _____

vii. Rubindium chloride injection (contains Rb-86) is used to determine _____ (7) _____ Blood flow.

viii. Benzil on treatment with urea gives _____ (8) _____ which is an anticonvulsant.

ix. Phenyl Malonamide on condensation with Form amide give _____ (9) _____

x. In the outer part of the _____ (10) _____ numerous ovoid Schizo-lysigenous oil glands are present in clove bud

xi. The brown seed of Nutmeg is surrounded by a crimson reticulate _____ (11) _____ stripped off and dried to form mace.

xii. Terbutaline sulphate is a _____ (12) _____ agonist

xiii. Salicylates and other anti-inflammatory drugs irreversibly inactivate the enzyme _____ (13) _____ responsible for the conversion of arachidonic acid to Prostaglandin.

xiv. The temperature above which cloudiness suddenly appears for non-ionic surfactant in Solution is known as _____ (14) _____

xv. Dragendorff's reagent used in testing alkaloids is chemically _____ (15) _____

xvi. IR absorption spectra of aldehydes, ketones, carboxylic acids and their derivatives contain one absorption peak common due to _____ (16) _____

xvii. When the molecular weight of substances is unknown, the intensity of absorption is expressed as _____ (17) _____

xviii. The frequency of rotating magnetic field and the frequency of the processing molecules become equal, they are said to be in _____ (18) _____

xix. A rod of silicone carbide 6-8 mm in diameter and 50 mm in length heated to a temperature of 1300°C gives _____ (19) _____ radiation.

xx. Osmotic pellets are extensively used as _____ (20) _____

3. State whether the following are TRUE or FALSE. If the statements are FALSE, correct them Give reasons and justify the statements in one or two sentences only.

- The chloro group in Chlorotetracycline is in the 8th position
- Ethacrynic acid is a high ceiling (loop) diuretic which inhibits electrolytic re-absorption in the thick ascending limb of the loop of Henle
- Pharmaceutical Pectin differs from commercial Pectin because it does not contain sugar or organic acid
- Antihistamines stimulates the metabolism of endogenous histamine
- Vitamin K appears to be compatible with mineral supplement such as Calcium and Iron.
- Hepatic clearance is the sum of hepatic metabolic clearance and the biliary clearance.
- The efficiency of a tumbling mixer is highly dependent on the speed of rotation.
- Liquid glucose is obtained by complete hydrolysis of starch.
- Ethylene oxide sterilization is suitable for rubber closures.
- Samples can be handled in the form of solid, liquid or gas in mass spectrometry.
- A photo cathode operates on the principle that electrons are emitted from certain material in direct proportion to the number of light quanta striking on the surface of the material
- Purine occurs free in nature.
- A guard column is used in HPLC to presaturate the mobile phase with the stationary phase.
- Balsams are resinous mixtures that contain large proportions of Benzoic acid, Cinnamic acid, Salicylic acid or esters of these acids.
- The State Pharmacy Council is established by the State Drugs Controller.

4. Match of the following

i. Match the following descriptions given in [P] to [U] with the products mentioned below

1. Agar [P] Chief carbohydrate from *Macrocystis pyrifera*
 2. Carrageenan [Q] Dried exudates from *Astragalus gummifer*
 3. Tragacanth [R] Closely related hydrocolloids from *Chondrus crispus*
 4. Algin [S] Hydrophilic colloid from *Gelidium cartilageneum*
 [T] Powdered endosperm of the seeds of *Cyamopsis tetragonolobus*
 [U] Carbohydrates from the Rhizomes of *Zingiber* spp.
 (a) 1-[S], 2-[R], 3-[Q], 4-[P]
 (b) 1-[P], 2-[R], 3-[S], 4-[T]
 (c) 1-[P], 2-[R], 3-[T], 4-[S]
 (d) 1-[P], 2-[Q], 3-[R], 4-[U]

ii. Listed are drugs 1 to 4. Their appropriate antihypertensive mechanisms are given in [P] to [U]. Match them correctly

1. Pindolol [P] Vasodilator
 2. Minoxidil [Q] Angiotensin converting enzyme inhibitor
 3. Captopril [R] Diuretic
 4. Amiloride [S] β -blocker
 [T] Centrally acting alpha adrenoceptor agonist
 [U] Potassium induction
 (a) 1-[S], 2-[Q], 3-[U], 4-[P]
 (b) 1-[P], 2-[R], 3-[S], 4-[T]
 (c) 1-[P], 2-[R], 3-[T], 4-[S]
 (d) 1-[S], 2-[P], 3-[Q], 4-[R]

iii. Listed below are substances which are assayed by organisms mentioned in [P] to [U] Match them correctly

1. Crystal Violet I.P. [P] *Pasteurella pestis*
 2. Ampicillin I.P. [Q] *Bacillus cerus*
 3. Plague Vaccine I.P. [R] *Macrococcus luteus*
 4. Rifampicin I.P. [S] *Staphylococcus aureus*
 [T] *Lactobacillus aureus*
 [U] *Bacillus subtilis*

(a) 1-[S], 2-[R], 3-[P], 4-[U]
 (b) 1-[P], 2-[R], 3-[S], 4-[T]
 (c) 1-[P], 2-[R], 3-[T], 4-[S]
 (d) 1-[P], 2-[Q], 3-[R], 4-[U]

iv. Important Psychoactive Phenothiazines listed below have the following side chains at position 10, which are given in [P] to [U]. Match them correctly

1. Chlorpromazine [P] 3- [4-methyl Piperazine 1-yl] Propyl
 2. Prochlorperazine [Q] 2- [1 methyl Piperid 2-yl] ethyl
 3. Thioridazine [R] 3- [4-(2 hydroxy ethyl) Piperazine 1-yl] propyl
 4. Perphenazine [S] Dimethyl amino ethyl
 [T] Dimethyl amino propyl
 [U] Dimethyl amino butyl

(a) 1-[S], 2-[Q], 3-[U], 4-[P]
 (b) 1-[T], 2-[P], 3-[Q], 4-[R]
 (c) 1-[P], 2-[R], 3-[T], 4-[S]
 (d) 1-[P], 2-[Q], 3-[R], 4-[U]

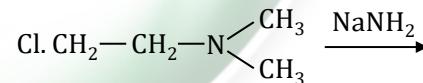
v. Listed below are the physical forms of the medicines which normally show the rheological properties given in [P] to [U]. Match them correctly

1. Viscous oils [P] Dilatant flow
 2. Gellies [Q] Plug flow
 3. Colloids [R] Newtonian flow
 4. Concentrated solid [S] Plastic flow suspension
 [T] Pseudoplastic flow
 [U] non-Newtonian flow
 (a) 1-[R], 2-[T], 3-[U], 4-[S]
 (b) 1-[P], 2-[R], 3-[S], 4-[T]
 (c) 1-[P], 2-[R], 3-[T], 4-[S]
 (d) 1-[P], 2-[Q], 3-[R], 4-[U]

PART - B

5. Complete the following reactions. Give the structural formula of the facts (In the answer book write the relevant products and their natural formula against each bold digit)

(i) Ethyl p-amino benzoate + 2 diethyl amino ethanol
 NaOC_2H_5 1. _____
 (ii) 2-Amino-5-chloro benzophenone + Ethyl glycinate
 Pyridine 2. _____ NaOCH_3 3. _____
 $(\text{CH}_3\text{CO})_2\text{SO}_2$
 (iii) 5-Methyl-iso oxazol 3 ethyl carboxylate + Benzyl hydrazine \longrightarrow 4. _____
 (iv) 10, 11 Dihydro-5H dibenz (b-f) azepine +
 $-\text{C}_2\text{H}_5\text{OH}$



6. Draw the structural formulae of the following

(i) A dibenzazepine derivative which is an anticonvulsant
 (ii) A piperidine derivative which is an opioid analgesic
 (iii) A naphthaquinone derivative which is an vitamin
 (v) Quinoline and pyrrole moieties in the structure, which is an anthelmintic
 (iv) A benzomorphan derivative which is an analgesic

7. List five important requirements in the formulation of controlled release parenterals. Each requirement should be written in one sentence only

8. (i) Ergot alkaloids are obtained on a commercial scale by two different methods. What are they
 (ii) What reagent is used for colorimetric assay of Ergot alkaloids

(iii) Given one specific test to detect the presence of
 (1) Anthraquinone glycosides in Senna leaves
 (2) Caffeine in tea leaves

9. Write only the equations for the various steps involved in the assay of
 (i) Ethosuximide I.P. (ii) Mephensin I.P.

10. Listed below are some important plant constituents. Give their chemical class and plant source
 (i) Camphor (ii) Digitoxigenin (iii) Hyoscine
 (iv) Xanthine (v) Pilocarpine

11. What reaction/test you would perform to prove the following
 (i) Nature and size of the ring of an alkaloid containing Pyridine nucleus
 (ii) Steroidal ring in Cholesterol
 (iii) Alpha, beta unsaturated lactone ring in Digitalis glycosides
 (iv) 7-CH₃ group in caffeine molecule

12. Starting from the following, outline the synthesis of
 (i) Mepyramine maleate from 4-Methoxy benzakichyde
 (ii) Cyproheptadine from 4-Chloro-1-methyl piperidine
 Use any other reagent if needed

13. Prepare a specimen label as per D and C. act and rules for 0.5 mg Busulphan tablets L.P 10x10 Tablets - Dose 2 to 4 mg daily

14. Draw the structures of the important metabolic products giving names of the transformation reaction
 (i) Tolbutamide (ii) Phenobarbital
 (iii) Chlorpromazine (iv) Imipramine
 (v) Salicylic acid

15. Draw the heterocyclic system present in the following drugs
 (i) Clonidine (ii) Cephalexin
 (iii) Methaqualone (iv) Thiotepa
 (v) Thiamine

16. Among the microscopical characteristics, the presence of different types of Calcium Oxalate crystals is an important diagnostic feature. Identify the correct type of Calcium Oxalate crystals present in the following drugs
 (i) Coca leaves
 (ii) Atropa belladonna leaves
 (iii) Mesophyls of Urginea maritima
 (v) Ailanthus glandulosa
 (iv) Datura stramonium leaves

17. Show the structural alteration in the compounds mentioned below and state what is the change in their activity
 (i) Introduction of F atom in the 5, position of Uracil
 (ii) Introduction of a mercrap to group in Hypoxanthine
 (iii) Conversion of ISONIAZID to 2, 2 dimethyl hydrazides

(iv) Preparation of a heterocyclic analog of Nicotinamide
 (v) One of the H atoms of the amino group of epinephrine is substituted by CH, CHCH₃,

18. Suggest specific mechanism of action of the following. Answer should be in one sentence only
 (i) Erythromycin (ii) Rifampin
 (iii) Oxacillin (iv) Nystatin
 (v) Cyclophosphamide

19. (i) Give the structural formulae of two stereoisomeric estradiols. Which is more potent
 (ii) What do you infer from the following observation?
 Answer in one sentence. In the lycopodium method for the determination of total length of fibres in a sample of Cinnamon bark powder gave 27 to 40 to 50 m per gm of air-dried powder
 (iii) Two different Senna leaf samples A and B gave the following values. Include the sample to the appropriate variety of Senna

Stomatal index	A	B
(Both surfaces)	17.1 to 18.7 to 20	11.4 to 12.2 to 13
Vein islet number	19 to 23	25 to 30

20. Give reasons for the following
 (i) A supporting electrolyte is added to a polarographic cell during analysis
 (ii) Conductivity of a solution is temperature dependent
 (iii) In the assay of alkali metal salts of Carboxylic acid Platinum crucibles are used and not porcelain crucibles
 (iv) Gas Chromatographic technique for pesticides, halogenated anesthetics etc., use an electron capture detector
 (v) Flavones on boiling with KOH and treating with FeCl₃, gives a violet colour

21. (i) In aerosol technology, certain specialized equipments are used to identify the factors mentioned below. Name them
 (a) To determine the particle size
 (b) To determine the flash point
 (c) To identify the propellant
 (ii) A powder has volume of 75 cm³ and a bulk volume of 125 cm³. Calculate its percentage porosity

22. (i) The following pharmaceutical aids have distinct disadvantages in their use. Name them in one or two sentences for each
 (a) Parabens (b) Cocoa butter (c) Polyamide
(ii) 10 ml ampoule of Potassium Chloride injection is available with labeled strength of 22.5 mEq/ml but actual requirement is 175 ml of 1 mEq/ml How will you reconstitute

23. Explain the following words and mention their significance in not more than two sentences for each
 (i) Caramalization (ii) Phase inversion temperature
 (iii) Solid fat index (iv) Ferrule (v) Case hardening

24. Give the principal function of the following equipments used in pharmaceutical industry

- (i) Ponman whistle (ii) Pycnometer
- (iii) Monsanto tester (iv) Breaking tester
- (v) Oscillating granulator

25. During the manufacturing of the tablets, the following defects were noticed. Give reasons for these defects in one sentence for each

- (i) Rat holing (ii) Blistering (iii) Hazing
- (iv) Picking (v) Double impression

26. Assing the main structural features of the compound

C_2H_2O from the following IR absorption data 1450
 cm^{-1} , 1265 cm^{-1} , 750 cm^{-1} , 1360 cm^{-1} and 1680 cm^{-1}

27. A Pharmaceutical formulation contains Zn, Mg and Cu ions. Suggest a suitable method to determine them without separation

28. What will be the adverse reactions, if the following drugs are administered together

- (i) Rifampin and Oral Contraceptive
- (ii) Tolbutamide and Sulphonamide
- (iii) Levodopa and Vitamin B
- (iv) Chloramphenicol and Phenobarbitone
- (v) Erythromycin and Carbamazepine

29. Give answers in one or two sentences only

- (i) Naloxone is N-allyl derivative of Oxymorphone. How does it exert its action
- (ii) What is positive ionotropic effect
- (iii) How does Verapamil, Nifedipine etc. act as a Calcium channel blocker
- (iv) How does Vinca alkaloids exert anticancer effects
- (v) What way the sulphonyl urea exerts their hypoglycemic effect

Answer Key

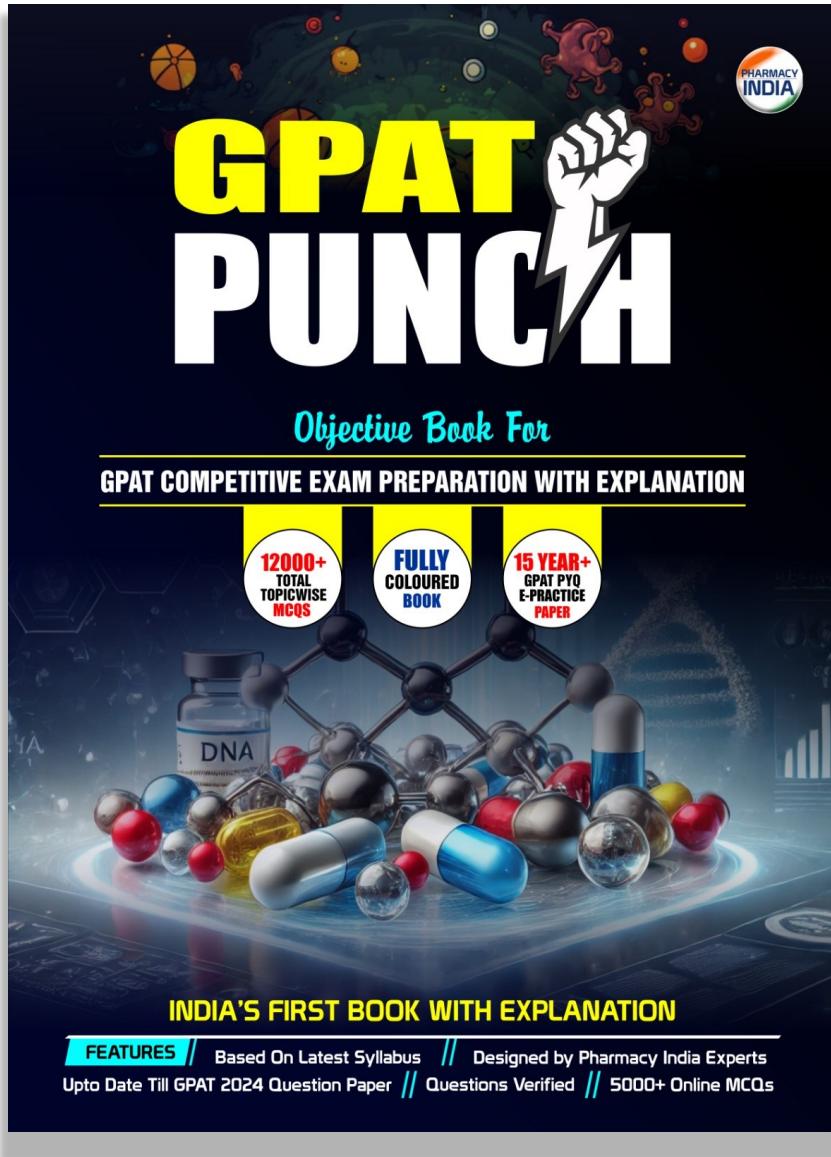
PART (SECTION - I)

i - a	ii - a	iii - d	iv - a	v - c	vi - c	vii - b	viii - c	ix - d	x - b
xi - b	xii - c	xiii - b	xiv - c	xv - a	xvi - b	xvii - c	xviii - c	xix - c	xx - a

PART (SECTION - II)

i - a	ii - d	iii - a	iv - b	v - a
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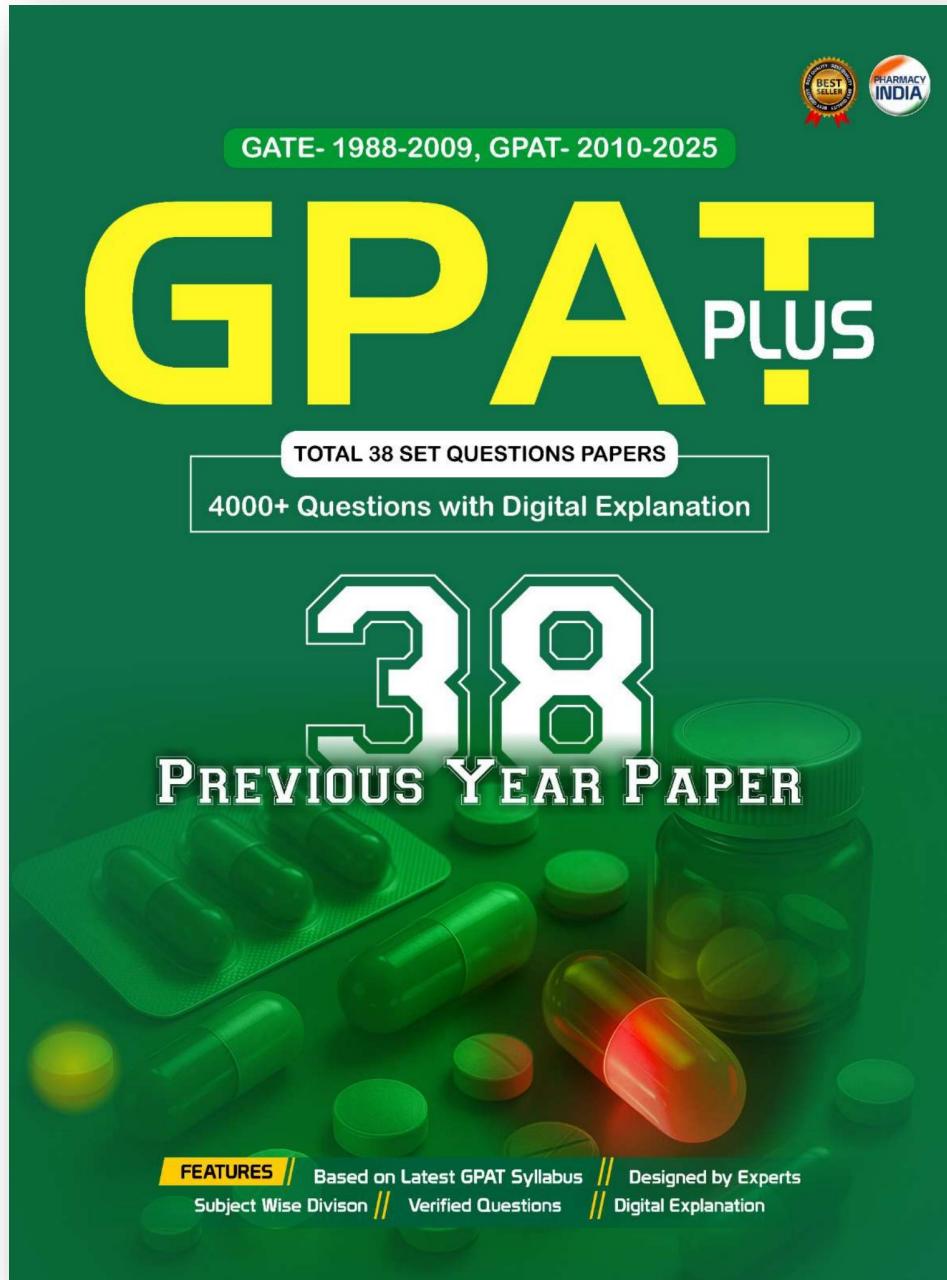
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