

## Pharmaceutics

1. Pitot tube measure \_\_\_\_\_ using a manometer
  - Surface area
  - Liquidity
  - Velocity head
  - Pressure head
2. What kind of substances can't permeate membranes by passive diffusion
  - Lipid-soluble
  - Non-ionized substances
  - Hydrophobic substances
  - Hydrophilic substances
3. Which among of the following variable is not a reason of errors during particle analysis by sieving method
  - Sieve loading
  - Intensity of agitation
  - Duration of agitation
  - Size of the particles
4. The time in minute at a specific temperature needed to kill a population of cell or spore is called
  - D-value
  - F-value
  - Z-value
  - None of these
5. Study the following statements  
**[P]** Fixatives in perfumes decrease rate of evaporation  
**[Q]** Astringent is main ingredient in antiperspirant  
**[R]** Melanin protects breakdown of bones  
**[S]** Hard water contains a lot of calcium and potassium salt  
 Choose the correct combination of statements
  - [P] and [Q] are correct
  - [R] and [S] are correct
  - [P] and [S] are correct
  - [Q] and [R] are correct
6. All are true EXCEPT
  - Soft soaps give emulsions with a pH in the basic range
  - Hard soaps form water-in-oil emulsions
  - Water-soluble polymers favour the formation of water-in-oil (w/o) emulsions
  - On the HLB system, lower numbers are assigned to lipophilic compounds while higher numbers are assigned to hydrophilic compounds.
7. Study conditions for long term stability testing of new drugs is
  - 30° C ± 2°C/ 65% RH ± 5% RH
  - 30°C ± 2°C/ 75% RH ± 5% RH
  - 40°C ± 2°C/ 65% RH ± 5% RH
  - 40° C ± 2°C/ 75% RH ± 5% RH
8. Myrij 52 is
  - Sodium oleate
  - Polyoxy ethylene Mono-stearate
  - Polyoxy lauryl ether
  - Polyoxy ethylene castor oil
9. Which among the following is the Henderson Hasselbach equation for a weak base and its salt
  - pH = pKa+Log([salt]/[base])
  - pH = pKa+Log([base]/[salt])
  - pH = pKb+Log([salt]/[base])
  - pH = pKb+ Log([base]/[salt])
10. Hydroxy ethyl carboxy methyl alkyl imidazolium hydroxide is an amphoteric surfactant which is known as
  - Macrogol
  - Miranol
  - Poly vinyl pyrrolidone
  - Povidone
11. The antibacterial activity of phenols is increased by
  - Increasing the pH
  - The presence of 10% vegetable oil phase
  - Increasing the temperature
  - The addition of a quaternary ammonium compound
12. Which of the following schedules includes shelf life of drugs
  - Schedule P
  - Schedule M
  - Schedule N
  - Schedule S
13. Which aerosol particles will be deposited in alveoli
  - >20 µm
  - <0.6 µm
  - 2 and 6 µm
  - 1-2 µm
14. How much can be the working revolution per minute (RPM) of the ball mill
  - 23-28 "D where D means the diameter of jar
  - Two times more than the critical revolution per minute
  - 42.3 D (D=diameter of the jar)
  - The average of critical RPM and the optional RPM
15. HEPA filter prevents the entry of microbes of size
  - Less than 0.3µ
  - Less than 0.1µ
  - More than 0.3µ
  - More than 0.1µ
16. The biological half-life of procaine in a patient was 35 minutes & Its V<sub>d</sub> was estimated to be 60L. The total clearance rate of procaine is
  - 1.188L/min
  - 0.115L/min
  - 11.5 L/min
  - 5.57L/min
17. Liposomes are used for all EXCEPT
  - Amphotericin-B
  - Doxorubicin
  - Propranolol
  - Vinceristine
18. The chelate, EDTA can described as what type of chelating ligand
  - Bidentate
  - Tetridentate
  - Hexadentate
  - Tridentate
19. How much duty has to charge for the Ayurveda preparations which can be consumed as alcoholic beverages per L.P. litre.
  - Rs. 2/-
  - Rs: 1/-
  - Rs. 5/-
  - Rs.30/-
20. Browne's tube is used
  - As a chemical indicator for sterilization
  - To check density
  - To measure viscosity
  - To store brown solutions

21. Which statistical tool is used for the bioequivalence study  
 (a) t- Test (b) Chi-squire test  
 (c) Null Hypothesis (d) ANOVA

22. Example for Desensitizing agents  
 (a) Zinc oxide (b) Strontium chloride  
 (c) Sodium fluoride (d) Calcium carbonate

23. For first order reactions the rate constant,  $k$ , has the units as  
 (a)  $\text{Ms}^{-1}$  (b)  $\text{M}^{-1} \text{s}^{-1}$  (c)  $\text{M}^{-2} \text{s}^{-1}$  (d)  $\text{s}^{-1}$

24. Sorensen apparatus is used for determination of  
 (a) Crystal growth (b) pH scale  
 (c) HLB value (d) Interarticular force

25. For the determination of dissolution profile of Micro-spheres, which is the best dissolution model we can apply  
 (a) Baker Lonsdale (b) Hixon-Crowell model  
 (c) Higuchi model (d) Noye's Whitney equation

26. Stokes Einstein equation to the series of  
 (a) Energy changes in Sedimentation suspension setting  
 (b) Sedimentation of suspension  
 (c) Diffusion coefficient  
 (d) Coefficient of energy consumption

27. 

Description	Parts of solvent required
1. Very Soluble	[P] 10-30
2. Spirangly soluble	[Q] 100-1000
3. Soluble	[R] < 1
4. Slightly Soluble	[S] 30-100

Match the following from specific group

(a) 1-[Q], 2-[R], 3-[P], 4-[S]  
 (b) 1-[P], 2-[Q], 3-[R], 4-[S]  
 (c) 1-[S], 2-[R], 3-[Q], 4-[P]  
 (d) 1-[R], 2-[S], 3-[P], 4-[Q]

28. Heckel's plot represents  
 (a) Extent of plastic and elastic deformation of material during compaction  
 (b) Force-time of force-displacement relationship  
 (c) Pressure-porosity (volume) relationship  
 (d) Stress relaxation measurements

29. Which of the following pharmacokinetics parameter is useful in representing the rate and extent of absorption of a drug  
 (a)  $C_{\text{max}}$  (b) Clearance  
 (c)  $t_{\text{max}}$  (d) Area under the curve

30. Stoke's formula for sedimentation velocity  $V$  is given by  
 (a) 
$$\frac{d^2(p_1-p_2)g}{18_n}$$
 (b) 
$$\frac{d^2(p_1+p_2)g}{18_n}$$
  
 (c) 
$$\frac{d^2(p_1+p_2)g}{9_n}$$
 (d) 
$$\frac{r^2(p_1-p_2)g}{18_n}$$

31. Dry Gum Method is also known as  
 (a) Continental method  
 (b) English method  
 (c) Bottle method  
 (d) Forbe's method

32. Which of the following is a clay type of emulsifying agent  
 (a) Xanthine (b) Vee gum  
 (c) Carbopol (d) CMC

33. Hot stage microscopy is a tool for the study of  
 (a) Pseudo polymorphism  
 (b) Particle size measurement  
 (c) Microbial contamination  
 (d) Compaction behaviour

34. Match the following

Drug	Influence of food on drug absorption
1. Aspirin	[P] Decreased
2. Erythromycin	[Q] Delayed
3. Griseofulvin	[R] Unaffected
4. Methyldopa	[S] Increased

(a) 1-[Q], 2-[P], 3-[S], 4-[R]  
 (b) 1-[P], 2-[Q], 3-[S], 4-[R]  
 (c) 1-[P], 2-[Q], 3-[R], 4-[S]  
 (d) 1-[P], 2-[R], 3-[S], 4-[Q]

35. Which marker is used for measuring the volume of plasma  
 (a) Deuterium oxide (b) Mannitol  
 (c) Evan blue (d) Inulin

## Pharmacology

36. What is the name of the tissues which helps in protection and support of the body  
 (a) Muscular tissue (b) Nervous tissue  
 (c) Connective tissue (d) Epithelial tissue

37. Which of the following drugs can cause Torsade de pointes  
 (a) Quinidine (b) Lignocaine  
 (c) Esmolol (d) Flecainide

38. Pulmonary fibrosis is the most common complication after treatment with  
 (a) 6- Mercaptopurine (b) Vincristine  
 (c) Bleomycin (d) Adriamycin

39. "Vaginal carcinoma in teenage female offspring's" abnormality observed in  
 (a) Methotrexate (b) Progestin  
 (c) Stilbesterol (Diethylstilbestrol) (d) Androgens

40. Wolff-chai koff effect is  
 (a) Reduced thyroxine synthesis by radiotherapy  
 (b) Reduced thyroxine synthesis by propylthiouracil  
 (c) Reduced thyroxine synthesis by iodides  
 (d) None of these

41. Mannitol is not used in  
 (a) Acute renal failure (b) Pulmonary edema  
 (c) Cerebral edema (d) All of these

42. Digitalis given to atrial flutter often produces  
 (a) Atrial fibrillation (b) Atrial tachycardia  
 (c) Atrial asystole (d) Atrial bigeminy

**43. What is the type of inhibition of acetylcholinesterase cause by organophosphates**  
 (a) Competitive, reversible  
 (b) Non-competitive, irreversible  
 (c) Uncompetitive, reversible  
 (d) Competitive, irreversible

**44. HIV may NOT respond to**  
 (a) Nucleoside analogues (b) Protease inhibitors  
 (c) Neuraminidase inhibitors (d) Reverse transcriptase

**45. Which of the following is a GLP-1 agonist**  
 (a) Sitagliptin (b) Pramlintide  
 (c) Exenatide (d) Eparlestat

**46. PDE-IV enzyme inhibitor agent used in the treatment of asthma**  
 (a) Roflumilast (b) Amoxicillin  
 (c) Salbutamol (d) Montelucast

**47. Which antifungal acts by non-competitive inhibition of squalene epoxidase thereby inhibition of ergosterol synthesis**  
 (a) Caspofungi (b) Amphotericin  
 (c) Terbinafine (d) Ketoconazole

**48. Which of the following is selective a, selective antagonist**  
 (a) Clonidine (b) Prazosin  
 (c) Phentolamine (d) Yohimbine

**49. Aromatase is an enzyme complex that is the target for several anticancer drugs. Which of the following anticancer drugs targets aromatase**  
 (a) Cyproterone acetate (b) Raloxifene  
 (c) Aminoglutethimide (d) Testosterone propionate

**50. Hepatic toxicity in paracetamol is caused due to an active metabolite. It is**  
 (a) N-acyl-p-benzoquinone mine (NAPQT)  
 (b) N-acetyl-p- benzoquinone mine (NAPQI)  
 (c) N-aryl-p-benzoquinone mine (NAPQI)  
 (d) N-alkyl-p-benzoquinone mine (NAPQI)

**51. Lorcaserin is used as**  
 (a) Anti-anxiety (b) Anti-smoking  
 (c) Anti-helminthic (d) Anti-obesity

**52. Which of the following is correctly matched**  
 (a) Dimercaprol Iron  
 (b) Calcium Di-sodium EDTA Arsenic  
 (c) Penicillamine Copper  
 (d) Deferoxamine Lead

**53. The agent that can only be given intravenously for heart failure is**  
 (a) Digoxin (b) Amiodarone  
 (c) Quinidine (d) Dobutamine

**54. Identify the GABA reuptake inhibitor**  
 (a) Pro gabide (b) Tiagabine  
 (c) Bicuculline (d) Baclofen

**55. The increase in concentration of second messengers (cAMP, cGMP, Ca<sup>2+</sup> etc) leads to**  
 (a) Inhibition of intracellular protein kinases and protein phosphorylation  
 (b) Protein kinases activation and protein phosphorylation

(c) Blocking of interaction between a receptor and an effector  
 (d) Antagonism with endogenous ligands

**56. The following may be used in erectile dysfunction EXPECT**  
 (a) Phenylephrine (b) Sildenafil  
 (c) Alprostadil (d) PGE, analogues

**57. Which part of internal ear receives sound waves in man**  
 (a) Cochlea (b) Lagena and utriculus  
 (c) Ampullae and utriculus (d) None of these

**58. All of the anabolic steroids are effective orally EXCEPT**  
 (a) Stanazolol (b) Methandienone  
 (c) Nandrolone (d) Oxymetholone

**59. Drug of choice for smoking cessation**  
 (a) Varenicline Which one of the (b) Duloxetine  
 (c) Acamprosate (d) Chlordiazepoxide

**60. following is used in the Ames test**  
 (a) Pseudomonas syringae  
 (b) Streptococcus aureus  
 (c) Pseudomonas aerogenosa  
 (d) Salmonella typhimurium

**61. The drug used during the preoperative preparation for surgical excision of pheochromocytoma is**  
 (a) Atenolol (b) Phenoxybenzamine  
 (c) Reserpine (d) Clonidine

**62. The pacemaker potential is due to**  
 (a) Fast Na<sup>+</sup> channel  
 (b) Decrease in K<sup>+</sup> permeability  
 (c) Slow Ca<sup>++</sup> channel  
 (d) Rapid repolarization

**63. N-acetyl cysteine is beneficial in acute paracetamol poisoning because**  
 (a) It reacts with paracetamol to form a nontoxic complex  
 (b) It inhibits the generation of the toxic metabolite of paracetamol  
 (c) It is a free radical scavenger  
 (d) It replenishes hepatic glutathione, stores which in turn binds the toxic metabolite of paracetamol

**64. An example of a major drug-drug interaction would be**  
 (a) Warfarin-Aspirin (b) Digoxin-Diltiazem  
 (c) Penicillin-Cephalexin (d) Hydrocodone-Codeine

**65. Hofmann elimination is seen in**  
 (a) Atracurium (b) Pancuronium  
 (c) d-TG (d) Vecuronium

**66. Paclitaxel acts on which phase of cell cycle**  
 (a) G<sub>1</sub> (b) S (c) G<sub>2</sub> (d) M

**67. Which of the following is species is being inactivated by the enzyme dipeptidyl peptidase**  
 (a) Oxytocin (b) Incretin  
 (c) Vasopressin (d) Glucagon

**68. \_\_\_\_\_ is the emergency contraceptive**  
 (a) Levonorgestrel  
 (b) Misoprostol  
 (c) Norethindrone  
 (d) Estradiol

69. The Vd for phenytoin is 70 L and half-life is 1.5 hours.  
**What is the total clearance of phenytoin**  
 (a) 34.32 L/hr (b) 32.34 L./hr  
 (c) 0.49 L/hr (d) 51.51 L/hr

70. **Not a hepatotoxic drug**  
 (a) Chlorpropamide (b) Allopurinol  
 (c) Streptomycin (d) Halothane

### Pharmacognosy

71. **Identify the mismatched pair**  
 (a) Hydathodes Piper betel (b) Aril lode- Cardamom  
 (c) Caruncle Belladonna (d) Strophiole- Colchicum

72. **In callus culture, roots can be induced by the supply of**  
 (a) Auxin and no cytokinin  
 (b) Higher concentration of auxin and lower concentration of cytokinin  
 (c) Higher concentration of cytokinin and lower concentration of auxin  
 (d) Auxin and cytokinin in equal proportions

73. **Atropine is**  
 (a) Optically active (b) Optically inactive  
 (c) Partially active (d) Partially inactive

74. **What are known as balsams**  
 (a) Resins dissolved in volatile oil  
 (b) A mixture of volatile oils with sesquiterpenes  
 (c) Solidified resin devoid of volatile terpenes  
 (d) Polysaccharide mixed with volatile oil

75. **Which of the following ergot alkaloids is water soluble and shows blue fluorescence**  
 (a) Ergosine (b) Ergotamine  
 (c) Ergocristine (d) Ergometrine

76. **Which of the following chemical constituents in fennel in a phenolic group**  
 (a) Anethole (b) Fenchone  
 (c) Eugenol (d) Carvonol

77. **Thalloequin test is for which of the following alkaloids**  
 (a) Barbaloin (b) Allopurinol  
 (c) Cinchona (d) Theobromine

78. **Identify the wrong statement regarding mucilage**  
 (a) They are normally sulphuric acid esters  
 (b) They are the normal products of the cell  
 (c) They are produced by injury to plant  
 (d) The ester group is a complex polysaccharide

79. **Which one of the following is NOT a protoplast fusion agent**  
 (a) Inactivated Sendai Virus (b) Ca<sup>++</sup> at alkaline pH  
 (c) Polyethylene glycol (d) Colchicine

80. **Isothiocyanate glycosides are prominent in family**  
 (a) Cruciferae (b) Liliaceae  
 (c) Rosaceae (d) Apocynaceae

81. **Types of fluorescence in UV light shown by Indian Rhubarb is**  
 (a) Deep violet (b) Blue (c) Yellow (d) Deep red

82. **"May apple" plant do possess**  
 (a) Anti-ulcer (b) Purgative  
 (c) Hypoglycaemic (d) Anti-cancer

83. **The monohydric alcohol present in Ashwagandha is**  
 (a) Anolide (b) Sorbitol  
 (c) Somital (d) Withanolide D

84. **Sunken stomata is the characteristic features of**  
 (a) Ephedra (b) Colchicum  
 (c) Lobelia (d) Senna

85. **Which of the following drug is sesquiterpene containing roots**  
 (a) Brahmi (b) Satavari (c) Valerian (d) Adusa

86. **Asafoetida + Nitric acid gives**  
 (a) Reddish-brown colour (b) Green colour  
 (c) Yellowish-orange colour (d) Blue fluorescence

### Pharmaceutical Chemistry

87. **Le Chatelier's principle states that increasing temperature favours a reaction that**  
 (a) Release energy as heat  
 (b) Requires energy as heat  
 (c) Involves a chemical catalyst  
 (d) Gluconeogenesis

88. **Which of the following has least polarity in bond**  
 (a) H-F (b) H-Cl (c) H-O (d) H-S

89. **Which one is di phenyl butyl piperidine derivative atypical antipsychotic agent**  
 (a) Molindone (b) Haloperidol  
 (c) Pimozide (d) Thioridazine

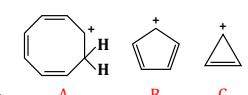
90. **p-nitro benzaldehyde and acetone to form 1-(4-nitro-phenyl-3-oxo-butene) is known**  
 (a) Michael condensation  
 (b) Claisen Schmidt condensation  
 (c) Alol condensation  
 (d) Claisen condensation

91. **Clinically used Labetalol is**  
 (a) S, S (b) R, R (c) R, S (d) S, R

92. **The positively polarized carbon atom of a carbonyl group acts as**  
 (a) An electrophile and Lewis's base  
 (b) A nucleophile and Lewis's acid  
 (c) A nucleophile and Lewis's base  
 (d) An electrophile and Lewis's acid

93. **Which of the following is an optically active compound**  
 (a) 1-Butanol (b) 1-Propanol  
 (c) 2-Chloro butane (d) 4-Hydroxy heptane

94. **Among the carbocation given below**  
 (a) A is homoaromatic, B is  
 Antiaromatic and C is aromatic  
 (b) A is Antiaromatic, B is  
 homoaromatic and C is aromatic  
 (c) A is aromatic, B is Antiaromatic and C is homoaromatic  
 (d) A is homoaromatic, B is homoaromatic and C is aromatic



95. The activity of which of the following drug is dependent upon a p-phenyl-N-alkyl Piperidine moiety  
 (a) Phenobarbital (b) Meperidine  
 (c) Imipramine (d) Chlorpromazine

96. The Selective H<sub>2</sub> - receptor antagonist agent should possess \_\_\_\_\_ for its minimum activity in its structure.  
 (a) Phenol (b) Imidazole  
 (c) Naphthalene (d) Quinoxaline

97. One of the following drug contains furan ring  
 (a) Cimetidine (b) Famotidine  
 (c) Ranitidine (d) Nizatidine

98. Which has most of acidic strength  
 (a) CH<sub>2</sub>-OH (b) CH<sub>3</sub>CH<sub>2</sub>-OH  
 (c) CH<sub>3</sub>-CH<sub>2</sub>-CH<sub>2</sub>-OH (d) CH<sub>3</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-OH

99. Which of the following is the name reaction of the given reaction  
 (a) Wolff's rearrangement  
 (b) Favorskii rearrangement  
 (c) Pinacol rearrangement  
 (d) Chisen rearrangement

100. Which of the following statements is true about ammonia and water  
 (a) Ammonia is more basic and more nucleophilic than water  
 (b) Ammonia is less basic and less nucleophilic than water  
 (c) Ammonia is more basic but less nucleophilic than water  
 (d) Ammonia is less basic but more nucleophilic than water

101. The reaction of Grignard reagent with aldehydes and ketones gives alcohols. This is known as  
 (a) Nucleophilic addition reaction  
 (b) Nucleophilic substitution reaction  
 (c) Electrophilic substitution reaction  
 (d) Electrophilic addition reaction

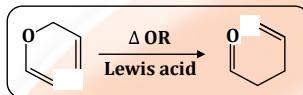
102. All the compounds have the molecular formula C<sub>4</sub>H<sub>8</sub>, the one which will show geometrical isomerism is  
 (a) H<sub>3</sub>C - CH=CH-CH<sub>3</sub> (b) H<sub>3</sub>C-C=CH<sub>2</sub>  
 (c) C<sub>2</sub>H<sub>5</sub>-CH=CH<sub>2</sub> (d) H<sub>3</sub>C-CH<sub>2</sub>-CH=CH<sub>2</sub>

103. What is the IUPAC name of the compound having the structural formula  
 (a) 3-Amino-2-hydroxypropanoic acid  
 (b) 2-Amino-propan-3-ol-1-oic acid  
 (c) Amino hydroxy propanoic acid  
 (d) 2-Amino-3-hydroxypropanoic acid

104. Electron pair of the pi bond is moved towards the attacking reagent due to  
 (a) +E effect (b) +1 effect (c) -E effect (d) -I effect

105. The rate of reaction is determined by slow step reaction. The step is called  
 (a) Reaction rate (b) Activation step  
 (c) Rate determining step (d) None of these

106. Which one of the following is acid stable  
 (a) Erythromycin stearate  
 (b) Erythromycin ethyl succinate  
 (c) Erythromycin estolate  
 (d) All of these



107. Thiotepa undergoes oxidative desulphuration forming an active cytotoxic metabolite  
 (a) Mon ethylene phosphor amide  
 (b) Diethylene phosphor amide  
 (c) Tri ethylene phosphor amide  
 (d) None of these

108. Any series of operations so carried out that at the end, the system is back to its initial state is called  
 (a) Boyle's cycle (b) Reversible process  
 (c) Adiabatic process (d) Cyclic process

### Pharmaceutical Analysis

109. On & scale the position of TMS signal is taken as  
 (a) 0.0 ppm (b) 0.2 ppm (c) 2.0 ppm (d) 10.0 ppm

110. Brockman activity Scale' is used in the characterization of  
 (a) Stationary phase (b) Mobile Phase  
 (c) Buffer System used (d) Column specification

111. Purity of water can be assessed by determining one of its following properties instrumentally  
 (a) Conductivity (b) pH  
 (c) Refractivity (d) Viscosity

112. Karplus curve is associated with which spectroscopy  
 (a) UV (b) Mass (c) FTIR (d) NMR

113. Phenobarbitone Sodium Tablets is officially assayed by  
 (a) Acidimetry (b) Precipitation titrations  
 (c) Redox titration (d) Gravimetric

114. If an organic compound does not absorb UV visible radiation it means compound does not contain \_\_\_\_\_  
 (a) Single bond (b) Sigma bond  
 (c) Conjugated double bond (d) Dative bond

115. Polarographic method of analysis to obtain individual amounts of Cu<sup>2+</sup> and Cd<sup>2+</sup> in a given mixture of the ions (Cu<sup>2+</sup> and Cd<sup>2+</sup>) is achieved by measuring them  
 (a) Half way potentials (b) Migration currents  
 (c) Decomposition potentials (d) Diffusion currents

116. The technique in which the change in weight of the substance is recorded as a function of temperature or time is  
 (a) Ignition (b) TGA (c) Incineration (d) DTA

117. Kjeldahl analysis is an acid-base titrimetric method for determining the amount of \_\_\_\_\_ in organic compounds  
 (a) Water (b) Carbon  
 (c) Nitrogen (d) Sulphur

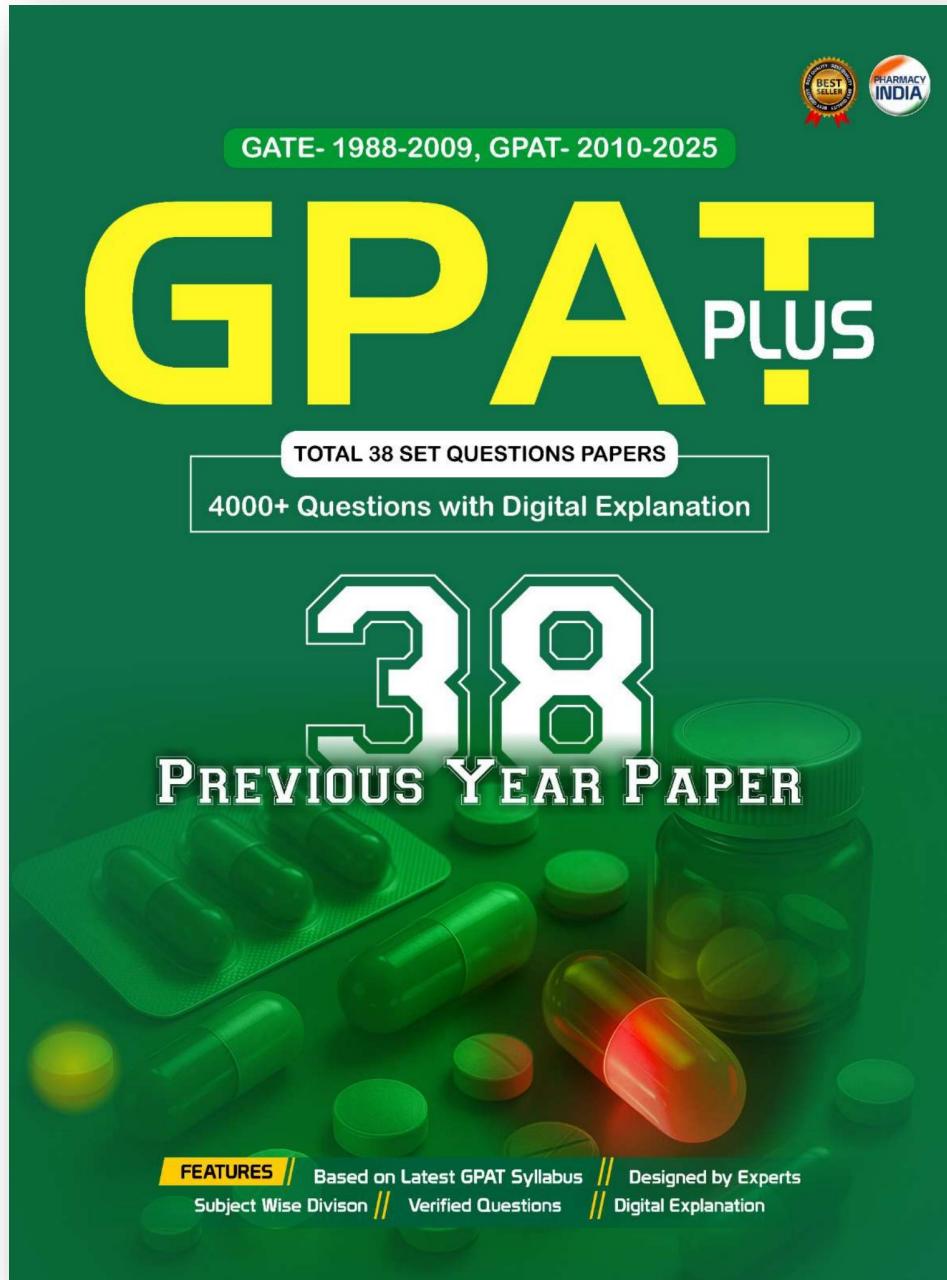
### Other Subjects

118. Which of the following criteria should be considered when reviewing a medication for addition to the hospital formulary  
 (a) The amount of samples provided to hospital physician  
 (b) Research funds donated to the hospital by the pharmaceutical company

## Answer Key

1-c	2-d	3-d	4-b	5-a	6-c	7-a	8-c	9-c	10-b
11-c	12-a	13-d	14-a	15-c	16-a	17-c	18-c	19-a	20-a
21-d	22-b	23-d	24-c	25-a	26-c	27-d	28-c	29-d	30-a
31-a	32-b	33-a	34-a	35-c	36-c	37-a	38-c	39-c	40-c
41-d	42-a	43-d	44-c	45-c	46-a	47-c	48-d	49-c	50-b
51-d	52-c	53-d	54-b	55-b	56-a	57-a	58-c	59-a	60-d
61-b	62-b	63-d	64-a	65-a	66-d	67-d	68-a	69-b	70-c
71-c	72-b	73-b	74-a	75-d	76-a	77-c	78-c	79-d	80-a
81-a	82-d	83-c	84-a	85-c	86-b	87-b	88-d	89-c	90-b
91-b	92-d	93-c	94-a	95-b	96-b	97-c	98-a	99-d	100-a
101-a	102-a	103-d	104-a	105-c	106-c	107-c	108-d	109-a	110-a
111-a	112-d	113-d	114-c	115-d	116-b	117-c	118-c	119-b	120-b
121-b	122-b	123-a	124-d	125-c					

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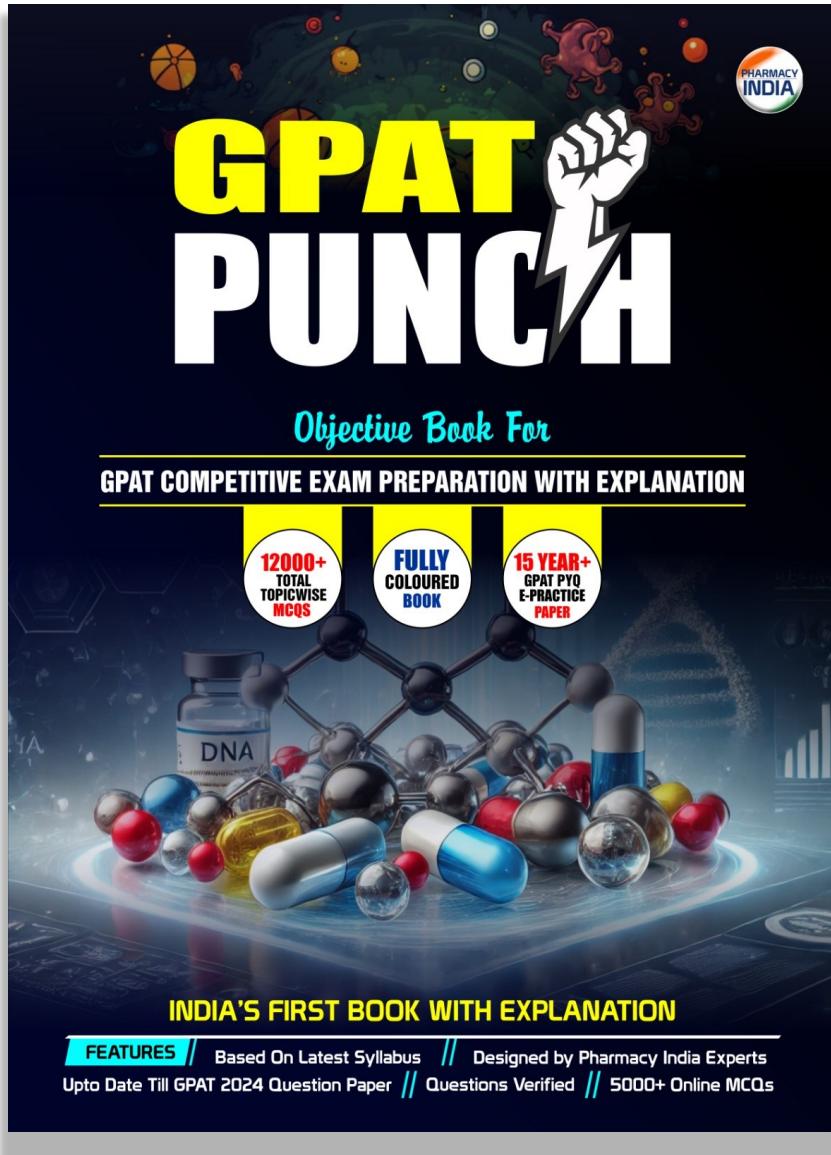
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